

# METHOD OF CLEANING AN OZONE GENERATOR

## FIELD OF THE INVENTION

5           The present invention is related to a method for cleaning electrical discharge ozone generators. More particularly, the present invention relates to circulating a warm fluid within the ozone generator to evaporate solid deposits of nitrogen oxides, including dinitrogen pentoxide, from the tubes and dielectrics of the generator.

## 10   BACKGROUND TO THE INVENTION

          Ozone ( $O_3$ ) is a strong oxidizing agent (2.07V) that is used as a disinfectant in various applications, such as wastewater treatment, cooling towers, air treatment, swimming pool cleaning, food processing, hydroponics, and meat processing. Ozone is particularly effective in aqueous environments. Ozone is, however, very reactive and cannot be stored for any significant period of time. As a result, ozone must be generated at the site where it is to be used. Two common means by which ozone is generated are by subjecting oxygen gas ( $O_2$ ) to ultraviolet radiation or to an electrical discharge.

20           One type of electrical discharge generator is an electrical barrier discharge ozone generator, commonly known as a silent discharge generator. One such generator is a corona discharge generator. Corona discharge generators are commonly used to generate ozone on a large scale. The basic principle of electrical discharge ozone generators is that a feed gas is fed through a high voltage electrical discharge field between two

electrodes. The oxygen is then ionized as it passes through the electrical field which will cause at least some oxygen to be converted to ozone. For corona discharge generators, the feed gas, usually dry air or oxygen, is subjected to coronal discharges created by high voltages between two electrodes, one of which is contained within a dielectric material.

- 5 In a tubular ozone generator, a dielectric is supported within a tube and a central cathode within the dielectric is subjected to a high voltage relative to an outer anode. The anode is often grounded. High voltage phenomena occur inside the dielectric envelope and induce a corona discharge field between the outside of the dielectric envelope (hereinafter referred to as “dielectric”) and the outer anode material. An oxygen-containing feed gas
- 10 is fed into this space and through this field, and the oxygen (O<sub>2</sub>) molecules are split to form atomic oxygen, which then reacts to form ozone.



- The quantity of ozone generated depends on several factors, such as for example
- 15 the voltage, the frequency of AC current, the gap between the dielectric and the cathode and the concentration of O<sub>2</sub> and other gases in the feed gas. The feed gas may be dry, clean air; dry, clean oxygen; or dry, clean oxygen containing small amounts of other relatively inert gases such as nitrogen (N<sub>2</sub>) or argon (Ar). It is important that dry feed gas be used, as water interferes with the reaction and also reacts with gases in the ozone gas
- 20 to create contaminants, most notably nitric acid (HNO<sub>3</sub>).

Much of the energy required for the reaction is lost as heat; therefore ozone generators should be cooled to operate more efficiently. One of the ways that cooling the

generator increases the efficiency of the generator is by causing fewer O<sub>3</sub> molecules to be lost due to decomposition or collision. A good description of ozone generating equipment can be found in U.S. Patent 4,954,321 by Jensen issued September 4, 1990, and also in "Ozone Technology and Equipment Design", Ozonia North America, USA  
5 1996, the information in, and contents of, both documents hereby being incorporated herein by reference.

Large amounts of ozone are not easily generated. For example, an ozone generation system employing an electrical discharge and which uses liquid oxygen at  
10 ≥99.5% purity, that has been vaporized and has had between 2 and 3% N<sub>2</sub> by weight and a certain amount of argon added, will typically produce 10 – 13% ozone by weight. Because of the relatively low rate of ozone generation, large plants may require several ozone generators to meet the demand for ozone. In turn, each generator may contain many anode/cathode/dielectric units.

As noted above, the amount of ozone generated depends on several factors, one of which is the amount of N<sub>2</sub> in the feed gas. When oxygen separated from air is used as a feed gas, nitrogen may be present. This is because of the method used to separate the oxygen from the air, e.g. vacuum or pressure swing adsorption or cryogenic separation.  
20 Nitrogen may also be present in the feed gas because it has been introduced to act as a catalyst. Nitrogen allows production of a higher ozone concentration or the reduction of the power consumed in generating the ozone. For example, large commercial ozone generators using pure oxygen generally create between 6 – 10% ozone by weight, instead

of the 10 -13% available when a small amount of  $N_2$  is added. It is therefore not desirable to remove all of the  $N_2$ .

Unfortunately, it has been discovered that the presence of nitrogen in the feed gas results in a solid residue, mainly composed of dinitrogen pentoxide ( $N_2O_5$ ), with some of it being deposited within the generator system, including on the tubes and the dielectrics. The residue may also contain other solid oxides of nitrogen ( $NO_x$ ). The oxide deposits on the support tubes and dielectrics and may eventually clog the passageway between the dielectric and the support tube.

Regular maintenance of ozone generators typically involves an inspection and repair of the electrical connections. However, because of the problems inherent in cleaning the generators described in greater detail below, opening the ozone generator to the atmosphere is avoided whenever possible. From time to time, however, ozone generators may require special or preventative maintenance. Such maintenance may be occasioned by failure of more than approximately 10% of the dielectrics or by deposits that clog the passages between the dielectrics and their support tubes in some systems.

Current methods of cleaning ozone generators consist of turning off the power supply and cooling water and purging the generator by circulating dry oxygen gas at room temperature through the system. The purging continues until the residual ozone has been removed from the inside of the generator for the safety of the workers. Thereafter the system is opened up to the atmosphere.

When the ozone generator is opened for regular maintenance, if it is opened for long enough, the water in the ambient air reacts with any residual solid nitrogen oxides to form nitric acid ( $\text{HNO}_3$ ). The reaction with  $\text{N}_2\text{O}_5$  for example, proceeds as follows:



Nitric acid is an oily, yellow residue, and any nitric acid in the generator needs to be removed.

The cleaning typically requires that all dielectrics and the tubes holding them be cleaned with a proper solvent. Generally, the dielectrics and tubes are removed from the system, cleaned with water and then with an industrial organic solvent such as acetone or a chlorinated organic solvent such as perchloroethylene, or methanol. This cleaning work is time consuming, and may require more than 14 days for an industrial scale ozone generator. In addition, removal and cleaning of the dielectrics will result in some breakage (perhaps 10%), thereby requiring their replacement. Finally, the chlorine containing solvents and the disposal of the contaminated cleaning solvents represent additional cost and safety issues that must be considered.

It is therefore desirable to have a less onerous cleaning method that would  
20 decrease the time and expense required for special maintenance of electrical discharge  
ozone generators, particularly large-scale corona discharge ozone generators.

## SUMMARY OF THE INVENTION

The current invention relates to a method of removing solid deposits of the oxides of nitrogen, including in particular dinitrogen pentoxide, in an ozone generator thereby  
5 avoiding the need to open the generator to atmosphere. If it is necessary to open the generator, to replace dielectrics for example, the inventive method will significantly reduce or eliminate the creation of nitric acid residue within the system. The inventive method can significantly reduce the maintenance time required from perhaps two to three weeks for each generator to perhaps as little as three days. In addition, damage to the  
10 dielectrics is minimized or eliminated, as is the need for solvents to remove the nitric acid. The method can provide significant costs savings in personnel time and materials.

A preferred embodiment of the inventive method uses warm gas circulation, preferably at 47-65°C, within the generator dielectric support tubes and warm water  
15 circulation in the shell section of the generator, preferably at 47-65°C. If the physical components of the generator can withstand temperatures above 65°C then the temperature of the gas can be increased well above 65°C, although this is not necessary to remove dinitrogen pentoxide, and heating the gas to a higher temperature may make the cleaning process more expensive. The fluid circulation within the system raises the  
20 temperature within the generator, and various solid oxides of nitrogen, which have boiling points less than the temperature of the circulated gas, including dinitrogen pentoxide which has a boiling point of about 47°C, are evaporated and thereafter

evacuated from the system by the gas stream. The temperature within the generator is sufficient to ensure that the deposits do not re-form within the ozone generator.

The progress of the cleaning can be monitored by bubbling a portion of the evacuated gas through a water trap and measuring the change in pH caused by the  $\text{HNO}_3$  formed by interaction of the  $\text{NO}_x$  and the water. Fluids are circulated within the generator until the pH of the water used as a reference is not appreciably lowered by the gas exiting the tubes of the generator.

If the need for maintenance was caused only by a build-up of  $\text{NO}_x$ , including in particular  $\text{N}_2\text{O}_5$ , the system is ready to return to production without requiring the generator to be opened to the atmosphere. Cleaning time and potential contamination are reduced. If the maintenance was required because of damaged dielectrics, when the system is opened to ambient air after being sufficiently cleaned, no nitric acid is formed. Only those dielectrics requiring replacement need be removed and replaced. Again, there are significant benefits in terms of both time and material savings.

In one aspect of the invention there is provided a method of cleaning an electrical discharge ozone generator comprising passing a warm cleaning gas between an inlet of the generator and an outlet of the generator to evaporate at least some of the  $\text{NO}_x$  deposited in the ozone generator.

In another aspect of the invention there is provided a method for removing solid deposits of  $\text{NO}_x$  from an ozone generator comprising a first and second electrode, the electrodes being spaced from each other and having a passageway therebetween. The solid deposits of  $\text{NO}_x$  are located within the passageway. The method comprises the step  
5 (i) of passing a warm cleaning gas through the passageway to evaporate the solid deposits of  $\text{NO}_x$  with boiling points equal to or less than  $65^\circ\text{C}$  which are deposited therein. The warm cleaning gas exiting the ozone generator is at a temperature sufficient to maintain the  $\text{NO}_x$  in a gaseous state until the  $\text{NO}_x$  exits the ozone generator.

10 In another aspect of the invention there is provided a method for removing solid deposits of  $\text{NO}_x$  from an ozone generator comprising a housing enclosing an interior having an inlet and an outlet and a pair of spaced electrodes mounted within the interior. The electrodes are spaced apart from each other. The solid deposits of  $\text{NO}_x$  are located within the interior. The method comprises the step of passing a warm cleaning gas  
15 through the interior from the inlet to the outlet to evaporate at least some of the  $\text{NO}_x$  deposited therein. The warm cleaning gas exits the ozone generator at a temperature sufficient to maintain the  $\text{NO}_x$  in a gaseous state until the  $\text{NO}_x$  exits the ozone generator.

20 In a further aspect of the invention there is provided a method for removing solid deposits of  $\text{NO}_x$  from an ozone generator comprising a housing and a plurality of support tubes mounted within the housing. The support tubes each support one or more dielectrics and each of the support tubes has an inner wall. A passageway is formed between the inner wall of the support tubes and the dielectrics. The passageway has solid deposits of

NO<sub>x</sub> therein. A support tube inlet is in flow communication with a support tube outlet through the passageway. The method comprises the step (i) of passing a warm cleaning gas through the passageway to evaporate at least some of the solid deposits of NO<sub>x</sub> which are deposited therein and carry at least some of the evaporated NO<sub>x</sub> from the ozone generator.

In another aspect of the invention there is provided a method for removing solid deposits of NO<sub>x</sub> from an ozone generator comprising an outer housing and a plurality of support tubes mounted within the housing. The support tubes each support one or more dielectrics and each of the support tubes has an inner wall and a passageway between the inner wall and the dielectrics. The passageway communicates between a support tube inlet and a support tube outlet. The housing has a shell that defines an interior surrounding the support tubes, the interior communicates between a shell inlet and a shell outlet. The method comprises step (i) of circulating a warm fluid within the shell and the concurrent step (ii) of evacuating the support tubes to remove the evaporated NO<sub>x</sub> with boiling points less than 65°C that had been deposited therein.

In a further aspect of the invention there is provided a method for removing solid deposits of NO<sub>x</sub> from an ozone generator comprising an outer housing and a plurality of support tubes mounted within the housing. The support tubes each support one or more dielectrics and each of the support tubes has an inner wall and a passageway between the inner wall and the one or more dielectrics. The passageway communicates between a support tube inlet and a support tube outlet. The housing has a shell that defines an

interior surrounding the support tubes. The interior communicates between a shell inlet and a shell outlet. The method comprises step (i) of circulating a cleaning gas within the support tubes and concurrent step (ii) of circulating a warm fluid within the shell to heat the cleaning gas, thereby removing the NO<sub>x</sub> with boiling points less than 65°C deposited therein. The temperature of the warm fluid is sufficient to ensure that the temperature of the cleaning gas exiting the ozone generator is sufficient to maintain the NO<sub>x</sub> in a gaseous state until the NO<sub>x</sub> exits said ozone generator.

In yet another aspect of the invention there is provided a method for removing dinitrogen pentoxide deposits from an ozone generator comprising an outer housing and a plurality of support tubes mounted within the housing. The support tubes each support one or more dielectrics and each support tube has an inner wall and a passageway between the inner wall and the dielectrics. The passageway communicates between a support tube inlet and a support tube outlet. A shell surrounds the support tubes, the shell defining an interior surrounding the support tubes. The interior communicates between a shell inlet and a shell outlet. The method comprises circulating a clean, dry mixture of oxygen, nitrogen and argon at 55°C - 60°C between the shell inlet and shell outlet; supplying the shell with warm water at 55°C - 60°C; diverting a portion of the gas exiting the support tubes to a liquid ring compressor; adding a neutralizing agent to the water in the compressor to maintain the pH in the liquid ring compressor at an approximately constant pH using an in-line process pH controller; and continuing the cleaning until the addition of neutralizing agent terminates as it is no longer required to maintain the constant pH.

Other aspects and features of the present invention will become apparent to those ordinarily skilled in the art upon review of the following description of specific, preferred embodiments of the invention in conjunction with the accompanying figures.

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## BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a schematic representation of a ozone generating system having an ozone generator which can be cleaned using a method in accordance with an embodiment of the invention.

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Fig. 2 is a cross-sectional view of part of an ozone generator, that can be cleaned using an embodiment of the inventive method.

Fig. 3 is a cross-sectional view at 3-3 of Fig. 2.

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## DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENTS

With reference to Figs. 1, 2 and 3, an ozone generator system 10 is illustrated which includes an ozone generator 14. It will be appreciated that Figures 2 and 3 show only part of an ozone generator. Typically such electrical discharge generators employ numerous dielectric support tubes through which oxygen is passed during the ozone generation process.

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In the Figures, generator 14 has a housing formed from a shell 14A. In this case, it is shown in Figures 1 and 2 with a jacket 14B surrounding shell 14A, although a jacket is not typically employed in commercial generators. Between shell 14A and jacket 14B is a jacket passageway 54. The jacket 14B has an inlet (not shown) which is in flow communication with an outlet (not shown) through passageway 54. Shell 14A houses a plurality of dielectric support tubes 28 which are mounted on supports within the interior 34 of shell 14A. Between the support tubes in interior 34 is an interior space 35 which surrounds the support tubes. Each support tube 28 houses one or more dielectrics 27. Each tube 28 has an inlet 25 which is in flow communication with an outlet 26 through a passageway 33 which is provided between the inner wall 50 of the support tube 28 and the outer wall 52 of the dielectrics 27. In operation to make ozone, ozone is created in the corona electrical discharge immediately outside or around dielectric 27. During the production of ozone, deposits 29 of nitrogen oxides, especially  $N_2O_5$ , may build up on the dielectric 27 and the inner wall support tube 28.

The system 10 also provides for cool water to flow through the interior space 35 of generator shell 14A around the tubes 28. During operation of system 10 to produce ozone, the flowing water will cool generator 14, thereby increasing the efficiency of ozone production. The flow of water 17A enters inlet (shown schematically as 18A) of shell 14A via conduit 36 and exits at outlet (shown schematically 18B). The water fills interior space 35, which as described above generally comprises the portion of interior 34 of the housing not filled by support tubes 28. However, when used in the inventive method of cleaning, the water flowing through interior space 35 is warm and may be

provided by water source 15 and heated at heater 16 or water may be provided by a source of warm water 32.

Also, during operation of system 10 to produce ozone, gas enters the ozone generator 14 at inlet 60 through conduit 31 (Figure 1), and then the gas flow divides within the generator 14 so that a portion of the total gas flow will flow through each of the passageways between the inlets and outlets of each support tube. At the outlets 26 of each support tube 28, the separate flows re-unite and then exit the generator at a common outlet 40. Providing a positive flow of gas is also preferred in practicing the inventive method. Thus during the cleaning process, a flow of gas 13A enters support tubes 28 at each inlet 25 and exits at each outlet 26. The cleaning gas may be provided by gas source 11 and heated at heater 12 or gas may be provided by gas source 30, which is already heated. After travelling through support tubes 28, the cleaning gas exits shell 14A via outlet 40 and may enter a water trap 21 via conduit 19. A portion of the cleaning gas that exits outlet 40 is diverted to outlet 20. Water trap 21 is provided with a source of reference water 23, and a pH monitor 22. Water exits water trap 21 through exit 24.

When ozone production is stopped at installation 10 to undertake special maintenance, occasioned, for example, by the support tubes 28 being plugged, such as by solid deposits of  $N_2O_5$  and perhaps other solid oxides of nitrogen in passageway 33 or by too many of the dielectrics 27 being damaged, the system is first purged of all  $O_3$ . This can be done, for example, by using the feed gas generally used to create ozone or by

using industrial grade oxygen. The gas is fed through support tubes 28 while no electric discharge is present.

With reference to Figs. 1-3, after purging the generator of ozone, in one embodiment of the invention, the inside of generator 14 where the dielectrics 27 and support tubes 28 are located is supplied through inlet 25 with a warm cleaning gas. The cleaning gas may be any dry, clean gas that is compatible with the ozone generator system, such as oxygen; nitrogen; a mixture of nitrogen and oxygen that may contain argon; or industrial grade helium, argon, air or possibly even carbon dioxide (CO<sub>2</sub>), although the latter will have the effect of lowering the pH to 7.0-8.0. It may be most convenient to utilize oxygen since that is the gas used in the ozone generating process. The cleaning gas must be dry, sufficiently contaminant-free and compatible with a system used for generating ozone, *i.e.* it should not detrimentally affect the physical system or interfere with production of ozone when the system is returned to production. Any such gas should be dry or substantially dry. In this embodiment, the cleaning gas evaporates and entrains the NO<sub>x</sub> 29 deposited on the dielectrics 27 and support tubes 28.

Herein, the term "cleaning gas", when used in this specification and claims includes any of the cleaning gases described thus far in the specification, and any other suitable gases.

In one preferred embodiment, the cleaning gas 11 is circulated into inlet 25 via conduit 31 and inlet 60 after heating at heat source 12 which may be any conventional

heat source, for example, a water bath, or an electrical or steam heater. In another preferred embodiment, a source of hot gas 30 may also be used. The gas enters the support tube 28 at inlet 25 at a temperature of preferably between 47 - 65°C and most preferably 55 - 60°C. The cleaning gas exits support tube 28 at outlet 26. Cleaning gas  
5 is circulated through the system 10 until the solid deposits of NO<sub>x</sub> that have boiling points of about 65°C or less, including in particular N<sub>2</sub>O<sub>5</sub>, have been substantially removed from the ozone generator.

The temperature of the warm gas 19A that exits tube 28, and later outlet 40, is  
10 preferably between 47 - 65°C to ensure that the evaporated NO<sub>x</sub> does not re-deposit as solids within the ozone generator. While not strictly necessary, it is good practice to ensure that NO<sub>x</sub> also does not re-deposit within conduit 19.

In the same preferred embodiments, while the gas is circulating in tube 28, water  
15 is circulated in the interior 34 of generator shell 14A. In one such preferred embodiment, the circulating water 15 is heated before entering the interior space 35 of shell 14A via conduit 36 and inlet 18A using a conventional heat source 16. As stated above, the heat source may, for example, be a water bath or an electrical or steam heater. In another embodiment, a source of warm water 32 is used. The water enters interior space 35 at  
20 inlet 18A at a temperature of between 29 - 65°C, preferably between 47 - 65°C and most preferably between 55 - 60°C and exits at outlet 18B at a temperature sufficient to ensure that, in combination with the cleaning gas temperature, the evaporated NO<sub>x</sub> remains in a gaseous state until it exits outlet 40.

While water circulation is preferred, it is not necessary. In most applications, gas circulation alone, should normally be sufficient to clean the system of the NO<sub>x</sub> deposits as long as the temperature reached inside the support tubes is sufficient to evaporate the NO<sub>x</sub> 29, as the gas passes over the NO<sub>x</sub> solids 29 and maintain the NO<sub>x</sub> in a gaseous state until the NO<sub>x</sub> exits the generator 14. It will also be appreciated that, at relatively high flow rates of cleaning gas, the NO<sub>x</sub> solids may be evaporated at a temperature that is significantly below their boiling points due to the vapor pressure effects.

In yet a further embodiment, the gas 13A or water 17A may be heated after entering the generator 14 by causing the generator itself to be heated. Such heating may take many forms, such as for example by applying a heat source directly to the outside of shell 14A or by circulating hot water or steam through a jacket 14B mounted on the outside of shell 14A, as long as the generator can withstand such heating.

Additionally, in a further embodiment of the invention there is no need to heat the cleaning gas directly if the temperature and the effect of the fluids circulating in interior space 35 or jacket 14B has a sufficient effect on heating the cleaning gas, that the cleaning gas can evaporate substantially all of the deposited NO<sub>x</sub> 29 and maintain the NO<sub>x</sub> in a gaseous state until it exits the generator 14 at outlet 40.

Additionally, it may also possible to remove the deposits of NO<sub>x</sub> 29 using water circulation only in generator shell 14A, as long as the temperature inside the support

tubes 28 of the generator 14 is sufficient to evaporate the NO<sub>x</sub> deposit 29. However, if this embodiment of the invention is used, a means for creating a flow of gas out of support tube 28, such as a vacuum pump will be required. In that case, the ozone generator used must be rated to withstand the physical stresses that may result.

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If warm water is used, it exits generator shell 14A at outlet 18B. The water exiting generator shell 14A has a temperature of preferably between 47 – 65°C. The water may be discharged in an environmentally safe manner or it may be re-circulated to inlet 32 if it is still warm although more likely it would be returned to inlet 15 and reheated prior to re-circulation through the generator.

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In a further embodiment of the invention, at least a portion of the cleaning gas exiting support tube 28 through outlet 26 enters a water trap 21 via conduit 19. The pH of the water in water trap 21 is monitored continuously or manually by a pH meter 22. The water trap 21 may be any one of several water containers such as a barrel, a tank or a liquid ring compressor, as long as it is sufficiently stable to withstand the expected gas flow into the water. The liquid ring compressor uses an elliptical liquid ring around an impeller to compress the ozone gas. As the ozone is compressed, it gives off heat, but this heat is absorbed by the ring of water. This water is continuously re-circulated through the compressor and through a heat exchanger to cool the water.

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The majority of the gas exits outlet 20 in front of water trap 21 to an approved scrubbing or capture system.

If NO<sub>x</sub> 29, including in particular N<sub>2</sub>O<sub>5</sub>, is present, it will react with reference water 23 flowing into the trap 21 and form nitric acid, thereby reducing the pH 24 of the water in the trap 21 below that of the incoming reference water 23.

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The reference water 23 is fed into the trap 21 continuously at a certain flow rate, which will depend on the particular system being cleaned, sufficient to record an appreciable pH change at the beginning of the cleaning cycle when the cleaning gas and warm water are at the proper temperature. The value of the appreciable pH change will depend on the pH monitoring system being used. The pH may also be monitored manually. The value of the pH drop can be approximately 3 pH units. The water trap 21 may be any size. The pH is allowed to vary and is monitored. When the monitored pH returns to the same pH as the incoming water, and stays constant, the cleaning will have been completed.

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In the preferred embodiment, the existing installation liquid ring compressor 21 and in-line pH control system, which includes a pH meter, are used to monitor the pH 24 of the water containing gas from conduit 19. When a compressor is used, it is not desirable to allow the pH to vary significantly as that might damage the compressor.

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Therefore, the method employed is that when the cleaning first begins, the pH of the reference water 23 in the liquid ring compressor 21 will start to drop. In response, the in-line pH control system starts to add a neutralizing agent, for example, trisodium phosphate (TSP), to maintain the pH at a substantially stable level. Thus the presence of

nitric acid can be ascertained by whether or not the neutralizing agent is continuing to be added. When the addition of the neutralizing agent stops, (*i.e.* no neutralizing agent is needed because the incoming gas no longer contains significant amounts of NO<sub>x</sub> and therefore no nitric acid is formed) the cleaning is complete.

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The circulation of warm gas 13A and/or warm water 17A is maintained until substantially all NO<sub>x</sub> 29 with boiling points less than 65°C, especially N<sub>2</sub>O<sub>5</sub>, have been removed from support tube 28 and dielectric 27 of the generator. Typically, this is when there is no longer an appreciable difference in pH between the pH 24 of the water exiting the trap 21 and the pH of the reference water 23 entering the trap 21, if no pH adjustment is applied.

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It will be appreciated by those skilled in the art, however, that pH monitoring, or more generally monitoring for the presence of NO<sub>x</sub> in the cleaning gas, is not necessary for the generator cleaning to be effectual. The cleaning may also be carried out for particular periods of time for which it is known that sufficient cleaning will have occurred, rather than monitoring. However, monitoring will clearly be a more accurate way of ensuring that the generator has been cleaned sufficiently.

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It will be appreciated by those skilled in the art that while water has been used to describe the above embodiments, other fluids including gases may be used within interior 34 and jacket 14B as long as they are compatible with the physical characteristics of the ozone generator being used.

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The aforementioned temperature ranges are influenced by the physical characteristics, including partially the boiling points of the NO<sub>x</sub> 29, and the physical limits of the particular generator used. The upper temperature limit of 65°C may be increased in generators that are constructed to withstand elevated temperatures. The upper limit of the temperature range is then dictated by other concerns such as safety concerns.

In the preferred embodiment, the O<sub>3</sub> is purged from the system before the generator cleaning process is started. However, if the ozone generating plant has available means of disposing of water contaminated with O<sub>3</sub>, the cleaning procedure can be commenced without first performing the O<sub>3</sub> purge.

The foregoing description is necessarily described with reference to the preferred embodiments of the inventive method applied to a particular ozone generator system but of course, the method may also be applied to other ozone generating apparatus. For example, the ozone generator may consist of spaced apart electrodes that are electrode plates and that have a passageway therebetween. While a plurality of embodiments of this invention has been illustrated in the accompanying drawings and described above, it will also be evident to those skilled in the art that changes and modifications may be made therein without departing from the invention. All such modifications or variations are considered to be within the scope of the invention as defined by the claims appended hereto.